



LBT & Image

”X-RAY SPECTROSCOPY OF BROMINE COMPOUNDS AND BIOMEDICAL APPLICATIONS”

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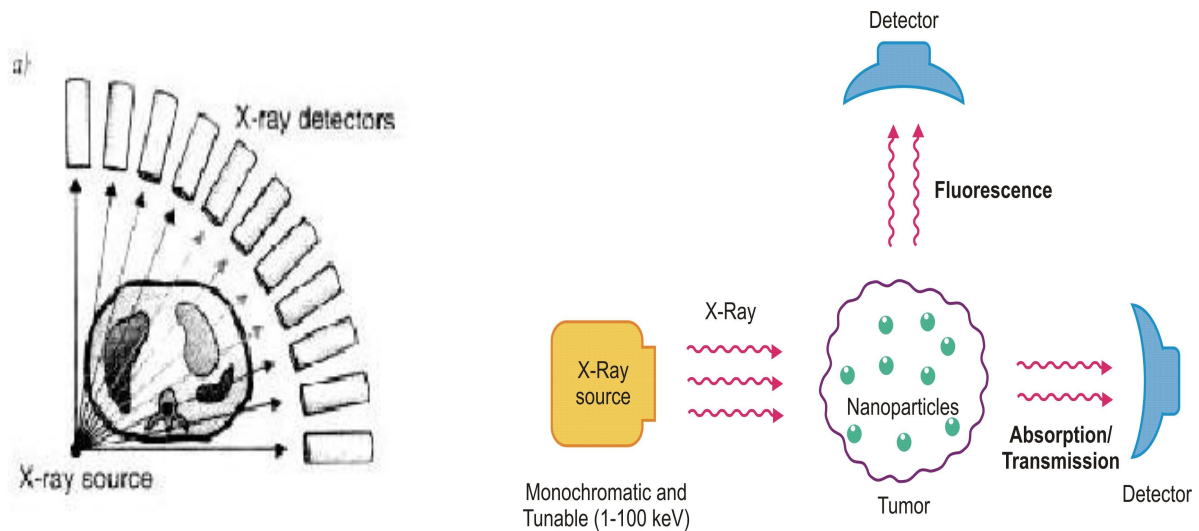
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NANO-BIO-SPECTROSCOPY



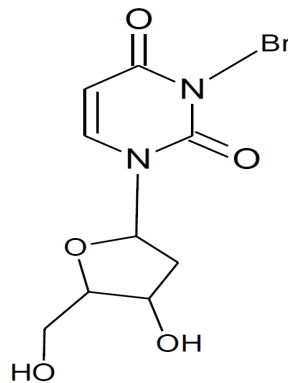
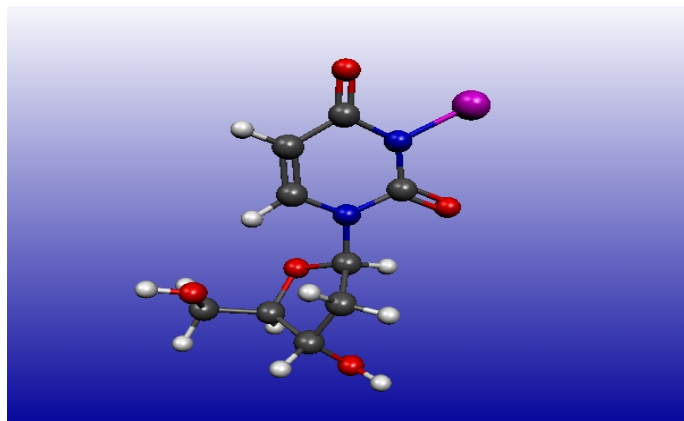
OBJECTIVE: Spectroscopy over Imaging

- Biomedical methods use intense and broadband high energy X-rays in radiation therapy and diagnostics (theranostics) to ensure sufficient tissue penetration (left Fig)
- To avoid damages incurred by these, nanobio-spectroscopy, **Resonant Theranostics**, aims to find narrow energy regions that correspond to *resonant or enhanced* absorption or emission (right fig)
- Spectroscopically targeted radiation - efficient with reduced exposure & provides detailed information

PRESENT: X-ray spectroscopy of the Br compound, bromodeoxyuridine (BUdR), widely used as radiological contrast agent in radiation imaging

X-RAY SPECTROSCOPY OF Br

- Tumors could be radiosensitized by halogenated thymidine analogs, e.g. BUdR (fig), for breakup of DNA of tumor cells



- The active element Br emits/ absorbs X-rays in narrow energy range: $\sim 12\text{-}14\text{ keV}$ ($1s \rightleftharpoons np$)
- Heavy elements absorb/emit X-rays at higher energies where biogenic elements (H,C,N,O,CHON) are transparent

RESULTS: Oscillator strengths (f) for Auger K-shell $1s$ - np transitions, photoabsorption cross sections (σ), photon absorption coefficients ($\kappa = \sigma/m$)

We show i) κ at resonant energies is much higher than the background and that of K-shell ionization

ii) Resonant energy bands lie *below* the K-shell ionization energy, $I_K \sim 14$ keV

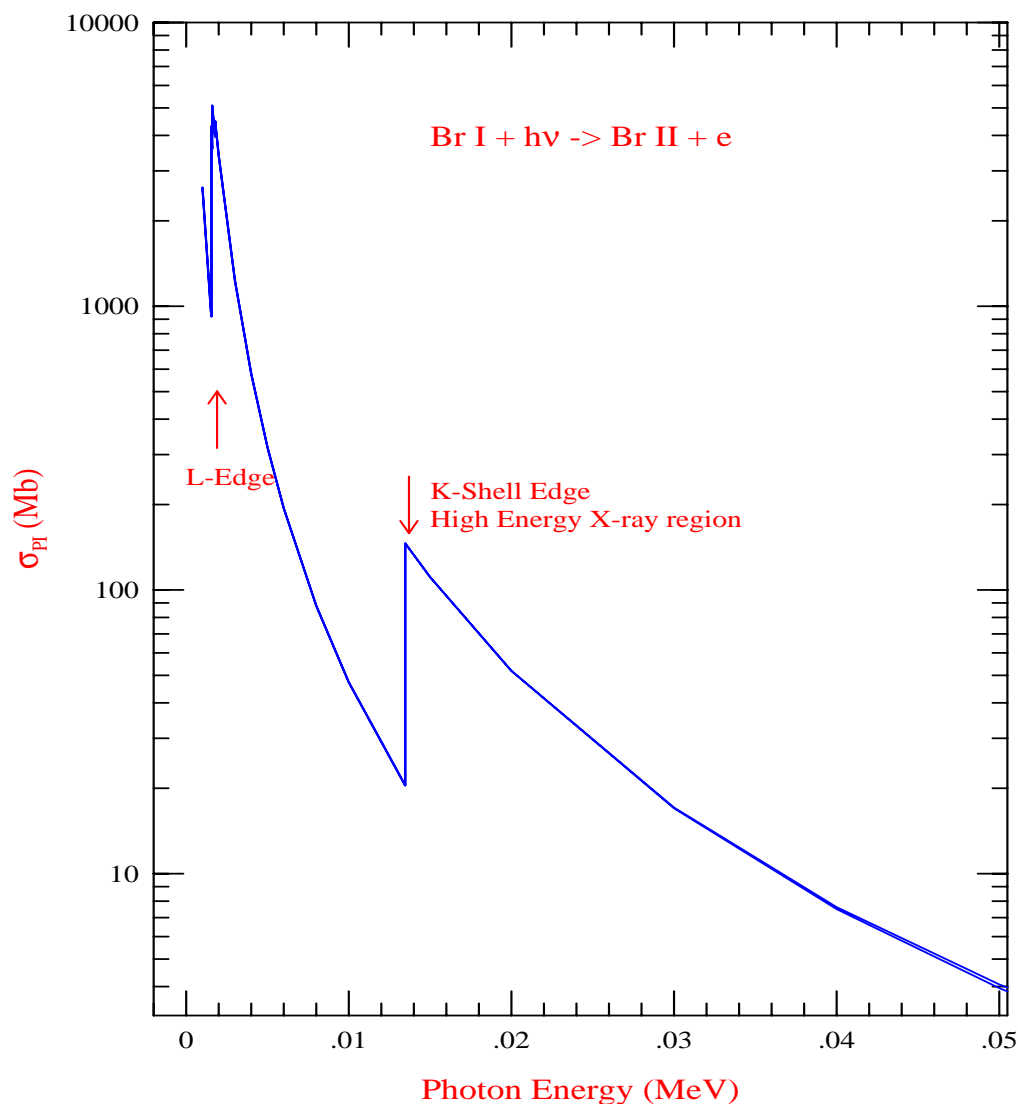
- These are contrary to the aim of research focus at the K-shell ionization
- Targeting these energy bands, Auger processes can be initiated to produce larger number of photons and electrons via photon fluorescence and electron ejections.

PHOTOABSORPTION: $X(\text{ion}) + h\nu \rightarrow X^+ + e$

X-RAY ABSORPTION BY Br AT K-SHELL EDGE

- Figure shows *background photoionization* (resonances excluded) cross sections σ_{PI} of Bromine
- Rises in σ_{PI} correspond to enhancements in ionization at various K, L, M (sub)-shells energies
- Br K-shell ionization edge ~ 14.0 keV. **Enhanced ionization \rightarrow enhanced emission of electrons**

Photoionization of Bromine

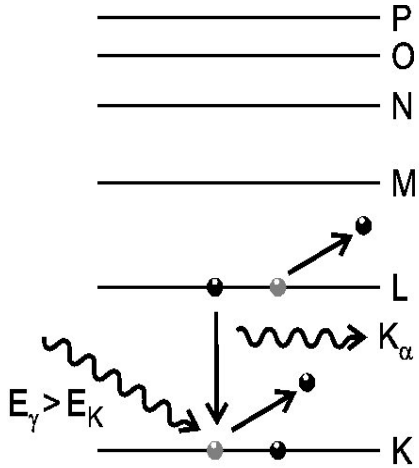


Auger Cascade: ELECTRON & PHOTON EMISSIONS

Auger Process: An electron from a higher level drops to fill a lower level vacancy, but emits a photon that can knock out another electron. This can lead to cascade as the vacancies move upward, more electrons and photons are emitted.

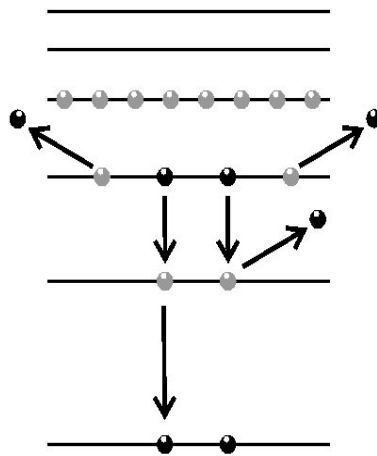
- Fig (i) Ionization by X-ray photons (E_X) ($>$ K-shell ionization energy E_K) leading to Auger process
- Fig (ii) Multiple electron vacancies due to successive Auger decays leading to emission of photons and electrons
- Single ionization of 1s electron can lead to ejection of 20 or more electrons in an ion with occupied O and P shells
- Fig (iii) Inverse to Auger - Resonant photo-excitation from 1s \rightarrow 2p (with L-shell vacancy) by an external monoenergetic X-ray source with intensity above our predicted critical flux

$$\Phi^c(\nu_{K\alpha}) = \frac{\sum_{n_i \geq 2} g_i A[n_i(S_i L_i J_i) \rightarrow 2(SLJ)]}{g_K B_{K\alpha}}. \quad (1)$$



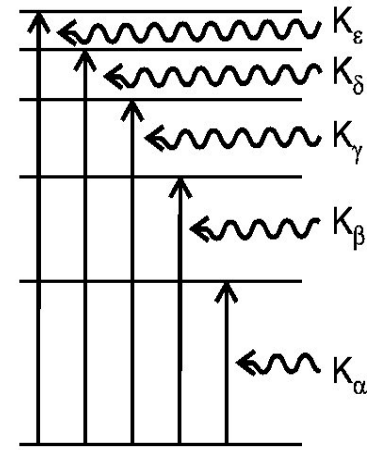
Ionization &
Auger Process

(i)



Electron Vacancies

(ii)



Resonant Excitation

(iii)

THEORY: RELATIVISTIC BREIT-PAULI APPROXIMATION

Relativistic effects: For a multi-electron system, the Breit-Pauli Hamiltonian is:

$$\mathbf{H}_{\text{BP}} = \mathbf{H}_{\text{NR}} + \mathbf{H}_{\text{mass}} + \mathbf{H}_{\text{Dar}} + \mathbf{H}_{\text{so}} + \frac{1}{2} \sum_{i \neq j}^N [\mathbf{g}_{ij}(\text{so} + \text{so}') + \mathbf{g}_{ij}(\text{ss}') + \mathbf{g}_{ij}(\text{css}') + \mathbf{g}_{ij}(\text{d}) + \mathbf{g}_{ij}(\text{oo}')] \quad (2)$$

where the non-relativistic Hamiltonian is

$$\mathbf{H}_{\text{NR}} \Psi = \left[\sum_{i=1}^N \left\{ -\nabla_i^2 - \frac{2Z}{r_i} + \sum_{j>i}^N \frac{2}{r_{ij}} \right\} \right] \Psi = \mathbf{E} \Psi \quad (3)$$

one-body corrections terms are

$\mathbf{H}_{\text{mass}} = -\frac{\alpha^2}{4} \sum_i \mathbf{p}_i^4$, $\mathbf{H}_{\text{Dar}} = \frac{\alpha^2}{4} \sum_i \nabla^2 \left(\frac{Z}{r_i} \right)$, $\mathbf{H}_{\text{so}} = \left[\frac{Ze^2 \hbar^2}{2m^2 c^2 r^3} \right]$ L.S the Breit interaction is

$$\mathbf{H}^{\text{B}} = \sum_{i>j} [\mathbf{g}_{ij}(\text{so} + \text{so}') + \mathbf{g}_{ij}(\text{ss}')] \quad (4)$$

Wave functions and energies are obtained solving:

$$\mathbf{H} \Psi = \mathbf{E} \Psi$$

- $\mathbf{E} < 0 \rightarrow$ Bound (e+ion) states Ψ_{B}
- $\mathbf{E} \geq 0 \rightarrow$ Continuum states Ψ_{F}

Transition Matrix elements with dipole operator $\mathbf{D} = \sum_i \mathbf{r}_i$,

$\langle \Psi_{\text{B}} || \mathbf{D} || \Psi_{\text{B}'} \rangle \rightarrow$ Radiative Excitation and Deexcitation

$\langle \Psi_{\text{B}} || \mathbf{D} || \Psi_{\text{F}} \rangle \rightarrow$ Photoionization and Recombination

The matrix element reduces to generalized line strength,

$$S = \left| \left\langle \Psi_f \left| \sum_{j=1}^{N+1} \mathbf{r}_j \right| \Psi_i \right\rangle \right|^2 \quad (5)$$

Oscillator Strength. Cross Section, Attenuation Coefficient

• The oscillator strength (f_{ij}) and radiative decay rate (A_{ji}) for the bound-bound transition are

$$f_{ij} = \left[\frac{E_{ji}}{3g_i} \right] S, \quad A_{ji}(\text{sec}^{-1}) = \left[0.8032 \times 10^{10} \frac{E_{ji}^3}{3g_j} \right] S \quad (6)$$

• The resonant structures of $K\alpha, K\beta, K\gamma, K\delta, K\eta$ complexes in photoionization cross sections σ_{PI} are obtained from Auger line strengths. For this, the resonant cross sections were convolved over a normalized gaussian function

$$\sigma_{K\alpha}(\nu; \mathbf{K} \rightarrow \mathbf{L}_i) = \frac{4\pi^2 a_o^2 \alpha}{3} \frac{E(\mathbf{K} - \mathbf{L}_i)}{g_k} S(\mathbf{K} - \mathbf{L}_i) \phi(\nu) = 4\pi^2 a_o^2 \alpha f_{KL_i} \phi(\nu) \quad (7)$$

where $\phi(\nu)$ the Gaussian profile with width ΔE ,

$$\phi(\nu) = \frac{1}{\sqrt{2\pi} \Delta E} \exp \left(-\frac{E^2}{2 \Delta E^2} \right)$$

• The mass attenuation coefficient per volume mass, κ is

$$\kappa_{\text{res}}(\nu; \mathbf{K}\alpha) = \frac{\sigma_{\text{tot}}}{\rho} = \frac{1}{u W_A} \frac{\sum_j w_j \sigma_{\text{res}}(\nu; \mathbf{K} - \mathbf{L}_{ji})}{\sum_j w_j} \quad (8)$$

where u is 1 amu, W_A = atomic weight, w_j =statistical weight

RESONANT $K_{\alpha,\beta}$ (1s-np) TRANSITIONS IN Br

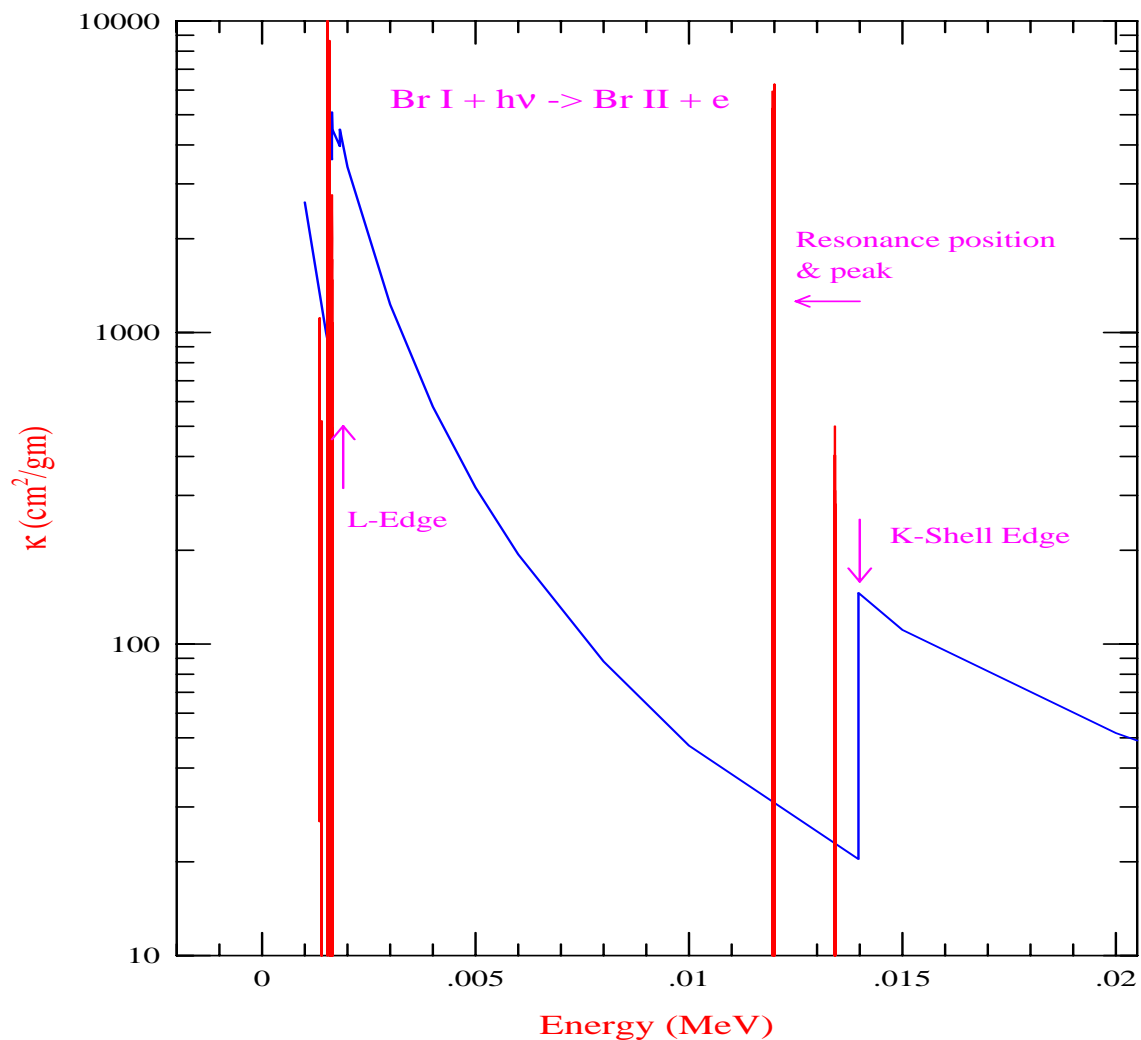
- The table presents data for resonant K -shell ($1s - np$) transitions in
Br II ($1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^4$)
- Auger transitions are initiated with K-shell ionization of Br I
- Table values are the averaged resonant energy, $E(K_{\alpha,\beta,\gamma})$, and the corresponding oscillator strength f and photoionization cross section, σ_{res}
- Each transition corresponds to a large number of fine structure transitions
- These resonance strengths provide the dominant contribution to photo-absorption, far more than K-shell ionization

Transition Array	# of Transitions	$E(K_\alpha)$ (keV)	f_{tot}	$\langle \sigma_{res}(K_\alpha) \rangle$ (Mb)
$1s2s^2 2p^5 \dots 4p^5 - 1s^2 2s^2 2p^5 \dots 4p^4$	30	8.809E+02	1.03E+00	8.31E+00
$1s2s^2 2p^5 \dots 4p^5 - \dots 3s^2 3p^5 \dots 4p^4$	30	9.864E+02	8.00E-02	6.45E-01
$1s2s^2 2p^6 \dots 4p^5 - 4p^4 4f$	59	9.993E+02	7.12E-09	5.74E-08
$1s2s^2 2p^6 \dots 4p^5 - \dots 4s^2 4p^5$	14	1.001E+03	1.34E-04	1.08E-03

X-ray Absorption Spectrum of Bromine

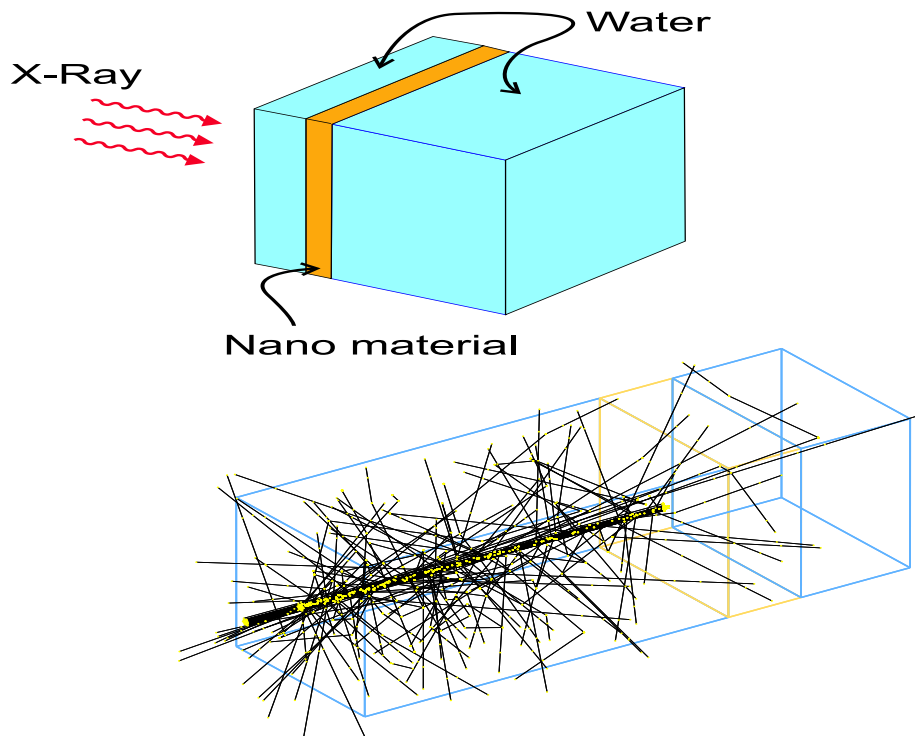
- **Figure:** X-ray mass photo-absorption coefficient $\kappa(cm^2/g)$
- The resonances, due to K \rightarrow L, M, N transitions, lie between E = 12 - 13.6 keV, below the K-edge
- Photo-absorption by the K-complex resonances exceed the background by more than an order
- Note, not studied before, that K-complex resonances are higher than the jump at K-shell ionization

Photo-absorption of Bromine



Monte Carlo Simulation for Resonant K_{α} X-Ray Absorption by Au Nanoparticles (Montenegro et al. 2009)

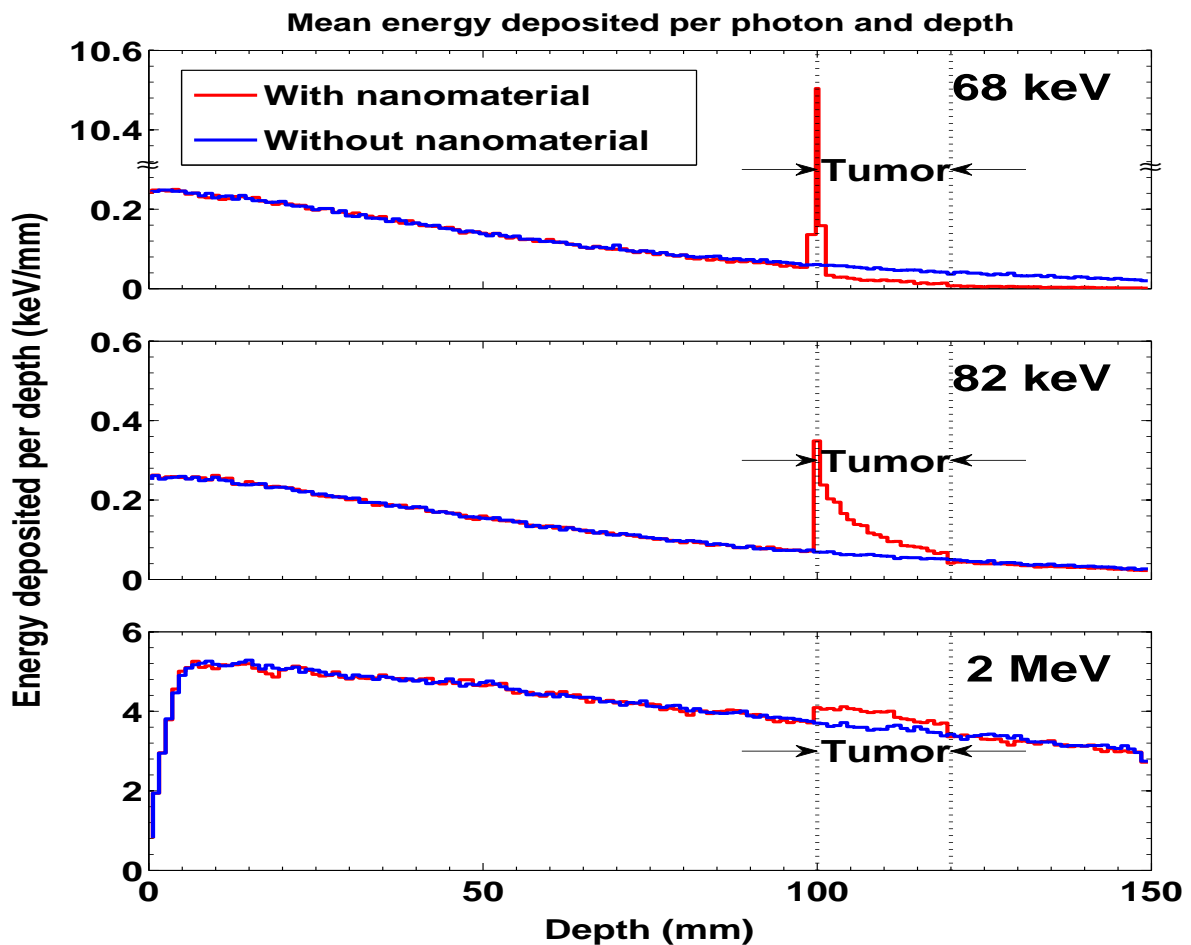
- **TOP:** Geometry of the experiment - the phantom (15 x 5 x 5 cm) models a tumor embedded with gold nanoparticles (golden section 2 cm) 10 cm inside normal tissue (blue section)
- **BOTTOM:** Simulation - gold nanoparticles at 5 mg/ml, X-ray beam at resonant energy ~ 68 keV
- **NOTE:** Because of Compton scattering only a few photons reach the region with gold nanoparticles



X-RAY ABSORPTION BY Au AT 68 keV, 82 keV, 2 MeV

Figure: X-ray energy deposited by depth in the phantom:
Red curve - with tumor in region 100 to 120 mm embedded with gold nanoparticles at 5 mg/ml, Blue curve - only water

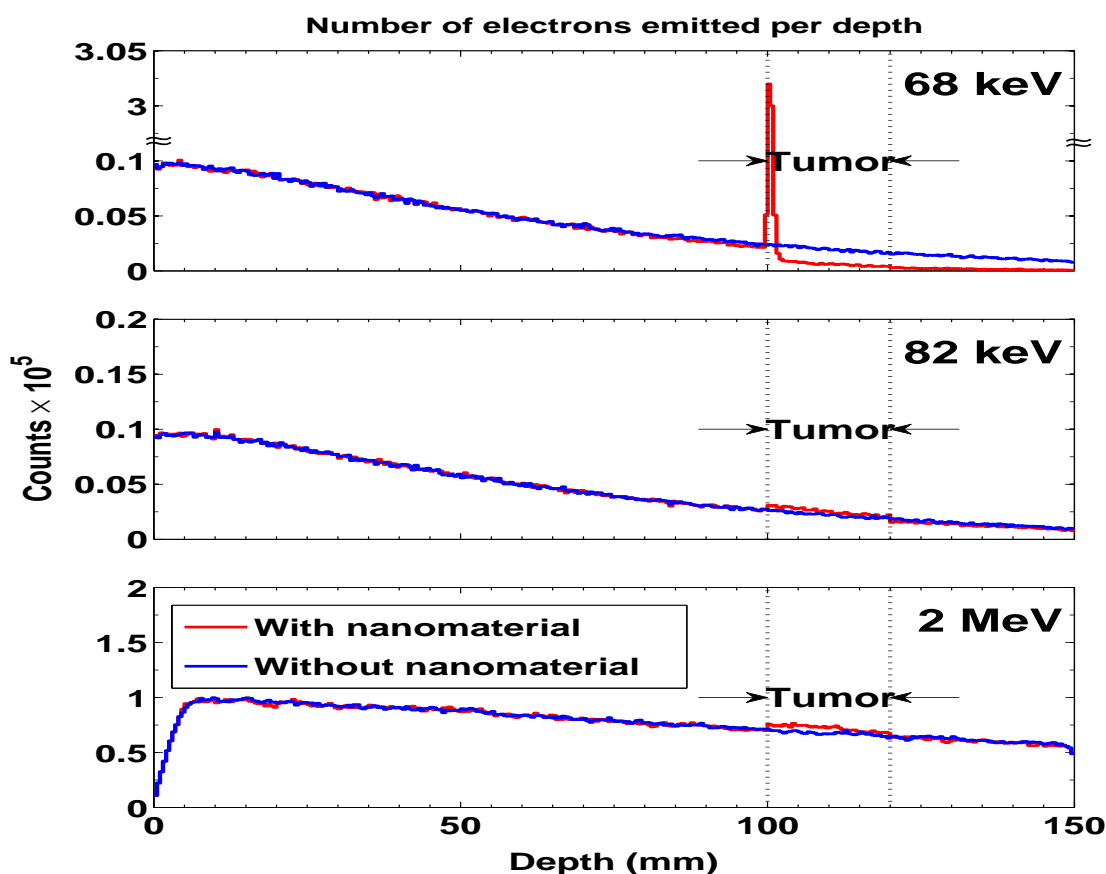
- **Top:** X-ray at 68 keV - averaged $K\alpha$ resonant energy
- **Middle:** 82 keV - just above K-edge ionization energy
- **Bottom:** 2 MeV - high energy common in clinical usage
- The presence of gold nanoparticles has increased the energy deposited at the tumor
- The highest absorption, by more than 25 times that at 82 keV, is at the resonant energy 68 keV (top panel)



ELECTRON PRODUCTIONS AT 68 keV, 82 keV, 2 MeV

Figure: Number of Auger electrons produced with depth following X-ray absorptions: Red curve - tumor embedded with gold nanoparticles at 5 mg/ml in region 100 to 120 mm, Blue curve - only water

- **Top:** X-ray at 68 keV - averaged $K\alpha$ resonant energy
- **Middle:** 82 keV - just above K-edge ionization energy
- **Bottom:** 2 MeV - high energy common in clinical usage
- A considerably large number of electrons, by more than an order of magnitude, were produced by 68 keV X-rays compared to those by 82 keV and 2 MeV



CONCLUSION

1. We present X-ray spectroscopy of the Br in BUrD where we predict resonant energies below the K-shell ionization threshold for enhanced X-ray absorption
2. We obtained Auger resonant probabilities and cross sections to obtain total mass attenuation coefficients with resonant cross sections
3. We find that the attenuation coefficients for X-ray absorptions at resonant energies are much larger, over an order of magnitude, higher over the background cross section as well as to that at K-edge threshold
4. The simulation for gold predicts enhanced X-ray absorption and follow-up enhanced electron and photon emissions at resonant energies.